



Structural studies of the extracellular polysaccharide from *Butyrivibrio fibrisolvens* strain CF3

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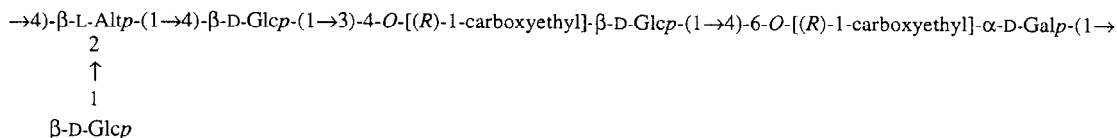
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Abstract

The structure of the *Butyrivibrio fibrisolvens* strain CF3 capsular polysaccharide has been investigated mainly by sugar and methylation analyses, Smith degradation, NMR spectroscopy, and mass spectrometry. The results indicate that the polysaccharide is composed of pentasaccharide repeating units having the following structure:



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1. Introduction

Butyrivibrio fibrisolvens is a strictly anaerobic bacterium most commonly isolated from the gastrointestinal tract of ruminant animals. Though principally involved in the catabolism of plant-derived polysaccharides, most strains of *B. fibrisolvens* also produce significant amounts of extracellular polysaccharides when grown in pure culture [1]. Previously, we have reported that these bacterially produced polymers contain an assortment of unusual monosaccharide constituents such as L-altrose [2], L-iduronic acid [3], several (1-carboxyethyl)-sugars [4], and others in a strain-specific manner [1].

We have also reported the detailed structures of the repeating units for the extracellular polysaccharides produced by *B. fibrisolvens* strains X6C61 [3] and 49 [5]. We continue these studies with other selected strains to define the structures of the repeating units of the major classes of extracellular polysaccharides produced by *B. fibrisolvens*. Previous studies with strain CF3, a bacterial isolate from the ovine cecum [6], have shown the presence of D-glucose, two unidentified (1-carboxyethyl)hexoses, and the unusual L-altrose, found for the first time in nature in this polysaccharide [2]. We now report further structural studies of this material.

2. Results and discussion

The crude capsular material was prepared as described [6] and then further fractionated by anion-exchange chromatography on DEAE-Sephrose to yield the pure polysaccharide. The ^1H NMR spectrum of the polysaccharide displayed broad peaks and contained, inter alia, signals from two methyl groups at δ 1.39 and 1.36 together corresponding to six protons, and from five anomeric protons at δ 5.23, 5.10, 4.80, 4.66, and 4.58. The ^{13}C NMR spectrum showed signals for five anomeric carbons at δ 101.77, 99.56 (two carbons), 104.07, and 103.08, corroborating the presence of a pentasaccharide repeating unit. The ^{13}C NMR spectrum also contained signals for two carbonyl carbons at δ 181.3 and 181.1, and for two methyl carbons at δ 19.3 and 18.4, indicating the presence of the two (1-carboxyethyl)-sugars reported previously [1,2].

The polysaccharide was carboxyl-reduced [7] using NaBH_4 as the reducing agent. A ^1H NMR spectrum of the product showed that the signals from the two methyl groups, originally at δ 1.39 and 1.36 in the unreduced polysaccharide, were shifted to 1.19 and 1.17, though not in a quantitative fashion. The reduction procedure was therefore repeated. A ^1H NMR spectrum of the re-reduced polysaccharide indicated

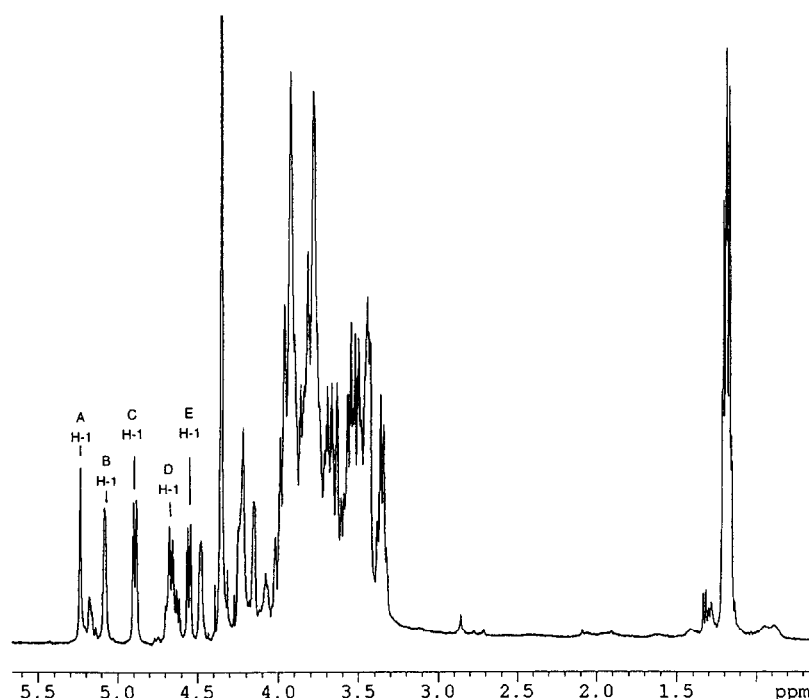


Fig. 1. 400-MHz ^1H NMR spectrum of the carboxyl-reduced extracellular polysaccharide from *Butyrivibrio fibrisolvens* strain CF3. For an explanation of A–E, see Table 1.

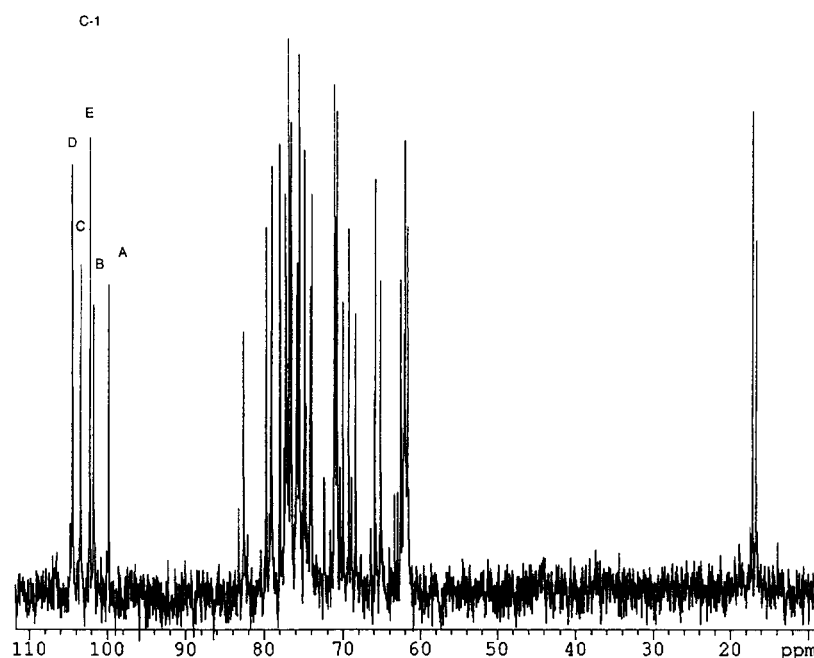


Fig. 2. 100.5-MHz ^{13}C NMR spectrum of the carboxyl-reduced extracellular polysaccharide from *Butyrivibrio fibrisolvens* strain CF3. For an explanation of A–E, see Table 1.

that the reduction was now essentially complete. The ^1H and ^{13}C NMR spectra (Figs. 1 and 2) and an HMQC spectrum of this product showed signals from five anomeric protons and carbons (Table 1), which corroborated the presence of a pentasaccharide repeating unit. The ^{13}C NMR spectrum also had signals at δ 17.13 and 16.69 corresponding to the two methyl group carbons in the two (2-hydroxy-1-methylethyl)hexoses.

Acid hydrolysis of the native polysaccharide, reduction with NaBH_4 and acetylation, gave the acetate of 1,6-anhydroaltrose and the alditol acetates of al-

trose and glucose in the relative proportions 0.6:0.4:2.5, confirming the ratio of one altrose to two glucose residues in the polysaccharide, as previously reported [2]. The L configuration of the altrose unit was reported earlier [2] and the D configuration of the glucose unit was now determined by GLC of the trimethylsilylated (+)-2-butyl glycosides [8].

The polysaccharide was carboxyl-reduced [7] with NaBD_4 , and, as in the case above, the procedure needed to be repeated in order to insure a more fully reduced product. The deuterium-reduced polysaccharide was hydrolyzed, the released sugars reduced with

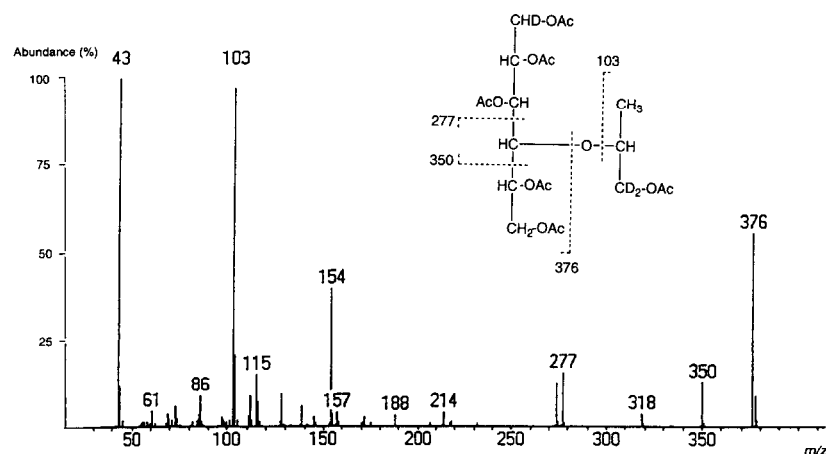


Fig. 3. Mass spectrum and the most abundant fragments for the alditol-1-d acetate of 4-O-[(R)-2-hydroxy-1-methylethyl]-2,2- d_2]-D-glucose.

Table 1
¹H and ¹³C NMR data obtained at 70 °C for the carboxyl-reduced polysaccharide from *Butyrivibrio fibrisolvens* strain CF3

Residue	Chemical shifts (δ)									
	H-1 C-1	(J _{H-1,H-2} , Hz) (J _{C-1,H-1} , Hz)	H-2 C-2	H-3 C-3	H-4 C-4	H-5 C-5	H-6a/H-6b C-6	2-Hydroxy-1-methylethyl		
								H-1 C-1	H-2a/H-2b C-2	1-Methyl C
→ 2,4)-β-L-Altp-(1 → (A)	5.23 99.77	(<2) (164)	4.15 77.10	4.48 68.31	3.92 75.48	3.90 73.80	3.78/3.96 62.50			
→ 4)-[6-O-R]-α-D-Galp-(1 → ^a (B)	5.08 101.69	(3.3) (171)	3.92 69.94	3.98 70.88	4.23 79.68	4.24 70.30	3.76/3.81 69.21	3.84 78.97	3.48/3.78 65.10	1.19 17.13
→ 4)-β-D-Glcp-(1 → (C)	4.89 103.31	(7.6) (166)	3.36 74.79	3.69 76.52	3.62 79.10	3.54 75.70	3.79/4.00 61.92			
→ 3)-[4-O-R]-β-D-Glcp-(1 → ^a (D)	4.66 104.41	(8.4) (161)	3.56 75.69	3.86 82.57	3.42 76.30	3.53 75.48	3.69/3.80 62.09	3.71 77.93	3.54/3.64 65.76	1.17 16.69
β-D-Glcp-(1 → (E)	4.55 102.14	(7.6) (161)	3.34 73.90	3.50 76.52	3.43 70.69	3.45 77.22	3.77/3.92 61.58			

^a R = (R)-2-hydroxy-1-methylethyl.

Table 2

Observed ^1H and ^{13}C chemical shifts ^a of the α and β forms of the isolated 4-*O*-[(*R*)-1-carboxyethyl]-D-glucose in D₂O at 70 °C

Anomer	Chemical shifts (δ)								
	H-1 C-1 $\Delta\delta_{\text{C}}$ ^b	H-2 C-2	H-3 C-3	H-4 C-4	H-5 C-5	H-6a C-6	H-6b	H-1' C-1'	H-2' C-2'
α	5.19	3.53	3.82	3.31	3.84	3.78	3.89	4.05	1.37
	92.08	71.95	72.31	78.72	68.24	60.37		77.44	19.12
	−0.91	−0.52	−1.47	8.01	−4.13	−1.47			
β	4.62	3.24	3.61	3.32	3.47	3.72	3.86	4.05	1.36
	95.79	74.50	74.92	78.78	75.34	60.27		77.44	19.65
	−1.05	−0.70	−1.84	8.07	−1.42	−1.57			

^a The chemical shifts of the carbonyl carbons were not determined.

^b ^{13}C chemical shift differences between the observed values and the literature data [12] for α - and β -D-glucose ($\delta_{\text{observed}} - \delta_{\text{literature}}$).

NaBD₄ and acetylated. The products were analyzed by GLC–MS, yielding the acetate of 1,6-anhydroaltrose and the alditol-1-*d* acetates of altrose, glucose, 6-*O*-(2-hydroxy-1-methylethyl-2,2-*d*₂)-galactose, and (2-hydroxy-1-methylethyl-2,2-*d*₂)-hexose in the ratio 1.2(1,6-anhydroaltrose and altrose):2.1:1.0:0.7, confirming the presence of one altrose, two glucose, one 6-*O*-(2-hydroxy-1-methylethyl-2,2-*d*₂)-galactose, and one (2-hydroxy-1-methylethyl-2,2-*d*₂)-hexose residue in the carboxyl-reduced polysaccharide.

The alditol acetate of 6-*O*-(2-hydroxy-1-methylethyl-2,2-*d*₂)-galactose was compared on GLC–MS with those prepared from the (*R*)- and (*S*)-forms of carboxyl-reduced synthetic methyl 6-*O*-(1-carboxyethyl)-D-galactopyranoside [9]. The retention times on GLC showed the component to be 6-*O*-[(*R*)-2-hydroxy-1-methylethyl-2,2-*d*₂]-D-galactose. This is the reduced form of 6-*O*-[(*R*)-1-carboxyethyl]-D-galactose, which consequently is the component in the native polysaccharide. This sugar has also been detected in the extracellular polysaccharides produced by the *B. fibrisolvens* strains X6C61 [3] and 49 [5].

The mass spectrum of the alditol acetate of the second (2-hydroxy-1-methylethyl-2,2-*d*₂)-hexose indicated a 4-*O*-[2-hydroxy-1-methylethyl-2,2-*d*₂]-hexose [10]. The spectrum and the origin of the principal fragments are depicted in Fig. 3.

To further investigate the structure of this component, the native polysaccharide was hydrolyzed, and the acidic sugars were first separated from the neutral sugars using an anion-exchange column. The mixture of the two (1-carboxyethyl)hexoses was then further

fractionated by chromatography on a cation-exchange column in the Ca²⁺-form [11]. ^1H NMR analysis of the fractions showed that one fraction contained the unknown pure (1-carboxyethyl)hexose. Treatment of an aliquot from the fraction containing the pure unknown (1-carboxyethyl)hexose with BBr₃, using a modification of the procedure described by Hough and Theobald [12] gave glucose, as determined by sugar analysis. The D configuration of the resulting glucose was established according to Gerwig et al. [8]. These data together with the data from the sugar analysis of the deuterium-reduced polysaccharide indicated the second (1-carboxyethyl)hexose to be 4-*O*-(1-carboxyethyl)-D-glucose. The ^1H and ^{13}C NMR signals of the isolated (1-carboxyethyl)hexose, assigned using different 2D NMR techniques (H,H-COSY, relay and double relay H,H-COSY and HMQC, see Table 2), were also consistent with a structure composed of a glucose unit with a 1-carboxyethyl substituent at O-4. This was evident from the downfield shift of the signal from C-4 (+8.0 and +8.1 ppm for the α and the β form, respectively) and the smaller upfield shift of the signals from the vicinal carbons (Table 2). To determine the absolute configuration of the 1-carboxyethyl substituent, the alditol acetate of 4-*O*-(2-hydroxy-1-methylethyl-2,2-*d*₂)-D-glucose obtained from the carboxyl-reduced polysaccharide was compared on GLC–MS with the alditol acetate of 4-*O*-[(*S*)-2-hydroxy-1-methylethyl]-D-glucose prepared from the carboxyl-reduced [7] extracellular polysaccharide from *Aerococcus viridans* var *homari* [13]. The different retention time of the model compound showed the polysaccharide constituent to be 4-*O*-

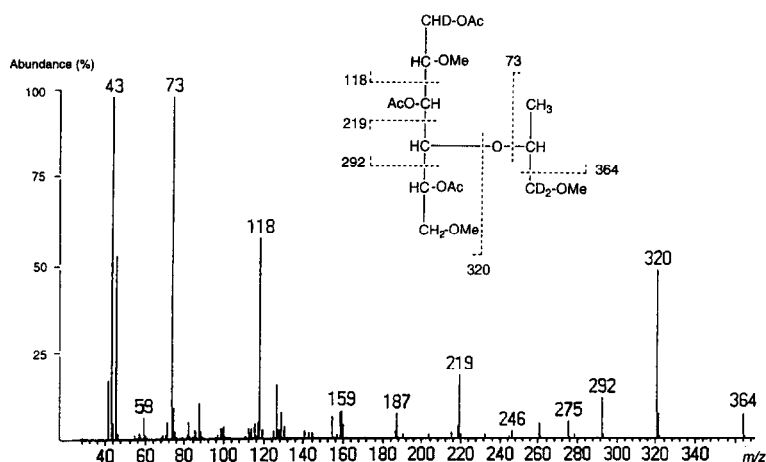


Fig. 4. Mass spectrum and the most abundant fragments for the alditol-1-*d* acetate of 2,6-di-*O*-methyl-4-*O*-[(*R*)-2-methoxy-1-methylethyl]-D-glucose.

[(*R*)-2-hydroxy-1-methylethyl]-D-glucose. This is the reduced form of 4-*O*-[(*R*)-1-carboxyethyl]-D-glucose, which consequently is the component in the native polysaccharide.

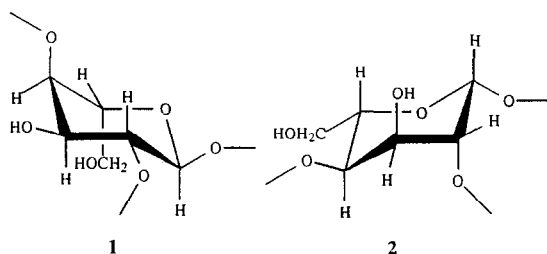
Methylation of the native polysaccharide was performed using sodium methylsulfinyl anion and CH₃I [14]. The resulting methyl esters of the permethylated polysaccharide were reduced with a solution of lithium triethylborodeuteride (Superdeuteride) [15]. The reduced, methylated polysaccharide was hydrolyzed, and the released sugars were reduced with NaBH₄ and acetylated. The products were analyzed by GLC-MS, which showed the alditol acetates of 2,3,4,6-tetra-*O*-methyl-D-glucose, 2,3,6-tri-*O*-methyl-D-glucose, 3,6-di-*O*-methyl-L-altrose, 2,3-di-*O*-methyl-6-*O*-(2-hydroxy-1-methylethyl-2,2-*d*₂)-D-galactose, and 2,6-di-*O*-methyl-4-*O*-(2-hydroxy-1-methylethyl-2,2-*d*₂)-D-glucose. However, the acetate of 2,6-di-*O*-methyl-4-*O*-(2-hydroxy-1-methylethyl-2,2-*d*₂)-D-glucitol gave few diagnostic fragments.

To confirm the above findings, a NaBH₄ carboxyl-reduced polysaccharide was methylated and hydrolyzed, and the released sugars were reduced with NaBD₄ and acetylated. Analysis of the products by GLC-MS gave the alditol-1-*d* acetates of 2,3,4,6-tetra-*O*-methyl-D-glucose, 2,3,6-tri-*O*-methyl-D-glucose, 3,6-di-*O*-methyl-L-altrose, 2,3-di-*O*-methyl-6-*O*-(2-methoxy-1-methylethyl)-D-galactose, and 2,6-di-*O*-methyl-4-*O*-(2-methoxy-1-methylethyl)-D-glucose. Fig. 4 shows the 70 eV EIMS spectrum of the acetate of 2,6-di-*O*-methyl-4-*O*-(2-methoxy-1-methylethyl)-D-glucitol-1-*d* together with the origin of the most abundant fragments. The above data indicated the repeating unit of the polysaccharide to be composed of a terminal D-glucose, a 4-substituted D-glu-

cose, a 2,4-disubstituted L-altrose, a 4-substituted 6-*O*-(1-carboxyethyl)-D-galactose, and a 3-substituted 4-*O*-(1-carboxyethyl)-D-glucose residue.

Information on the respective sugar residues and the anomeric configurations could be deduced from the ¹H and ¹³C NMR data of the carboxyl-reduced polysaccharide (Table 1). Using different 1D and 2D experiments, H,H-COSY, relay and double relay H,H-COSY, the spin-systems attributable to each anomeric proton could be assigned and the *J*_{1,2} values determined. HMQC experiments allowed then the assignment of the corresponding carbon signals and the ¹*J*_{C,H} values. The substitution positions of the residues were revealed by the high chemical shifts of the substituted carbon signals. From the pattern of several cross-peaks in the phase-sensitive COSY spectrum, the size of the coupling constants from the coupling between ring protons could be estimated. This information together with published chemical shift data for the sugars [3,5,16,17] allowed the assignment of the different spin-systems to specific sugar residues. The NMR data also showed that the 4-substituted 6-*O*-(1-carboxyethyl)-D-galactose was an α-pyranoside whereas the two D-glucose and the 3-substituted 4-*O*-(1-carboxyethyl)-D-glucose residues were β-pyranosides. The determination of the anomeric configuration of L-altrose required some additional considerations. Free altrose in solution is in equilibrium between the ¹C₄ and ⁴C₁ chair conformations [18], and as the former would have an axial hydroxyl substituent in the 2-position, the ³*J*_{H-1,H-2} value is not a reliable indicator of the anomeric configuration. The observed ¹*J*_{C,H} value (164 Hz) indicates that H-1 of L-altrose in the carboxyl-reduced polysaccharide is in an axial position [19]. This

fact indicates that the L-altrose residue must be in either the α configuration in a 4C_1 conformation (1) or in the β configuration in a 1C_4 conformation (2).

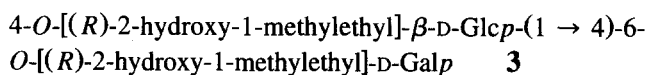


The small ${}^3J_{H-1,H-2}$ value found (< 2 Hz) precludes the first possibility (1), since in that case H-1 is in a trans-diaxial relationship with H-2, and the expected value should be 7–8 Hz. The measured ${}^3J_{H-1,H-2}$ value is instead in good agreement with the value expected for an axial-equatorial relationship, as shown in 2. The above findings were corroborated by the 3J values between vicinal ring protons in the L-altrose residue (< 2 , 4.3, and < 2 Hz, for ${}^3J_{H-1,H-2}$, ${}^3J_{H-2,H-3}$, and ${}^3J_{H-3,H-4}$, respectively) measured from a resolution-enhanced 1H NMR spectrum, the cross-peak patterns from a phase-sensitive H,H-COSY spectrum and from NOE difference experiments irradiating the H-1, H-2, and H-3 signals. The NOE connectivity between H-1 (δ 5.23) and H-5 (δ 3.35) observed in a NOESY experiment indicated a 1,3-diaxial relationship for these protons, and thus the L-altrose residue in the reduced polysaccharide preferentially adopts the 1C_4 conformation (2).

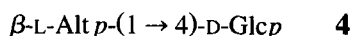
In order to deduce the sequence of the sugar residues, the carboxyl-reduced polysaccharide was subjected to a Smith degradation [20]. It was first oxidized with sodium periodate, then reduced with sodium borohydride, and the resulting polymer was isolated by dialysis. A 1H NMR spectrum of the product indicated incomplete oxidation, thus the procedure was repeated. Sugar analysis of the final product gave L-altrose, 1,6-anhydro-L-altrose, 4-*O*-[(*R*)-2-hydroxy-1-methylethyl]-D-glucose, erythritol, and 1-*O*-(2-hydroxy-1-methylethyl)-threitol identified by GLC–MS as their alditol acetates. The identity of 1-*O*-(2-hydroxy-1-methylethyl)-threitol was confirmed by comparison with an authentic sample prepared by Smith degradation of the reduced *B. fibrisolvens* strain 49 capsular polysaccharide [5]. The 1H NMR spectrum of the polymer showed only two anomeric proton signals (indicating two intact sugar residues), in agreement with the sugar analysis. From the chemical shifts and coupling constants (δ 5.19,

$J_{1,2} < 2$ Hz and δ 4.64, $J_{1,2}$ 7.6 Hz) of the signals, they could be assigned to the 2,4-disubstituted β -L-altrose residue and the 3-substituted 4-*O*-[(*R*)-2-hydroxy-1-methylethyl]- β -D-glucose residue, respectively.

Mild acid hydrolysis (0.5 M trifluoroacetic acid at room temperature) of the polymer mentioned above followed by separation of the hydrolysis products by gel filtration on a column of Bio-Gel P-2 yielded two saccharides. Sugar analysis of the first eluting saccharide gave equimolar amounts of 4-*O*-[(*R*)-2-hydroxy-1-methylethyl]-D-glucose and 1-*O*-[2-hydroxy-1-methylethyl]-D-threitol. The 1H NMR spectrum showed one anomeric proton signal at δ 4.59 ($J_{1,2}$ 7.9 Hz), confirming the β configuration of the 4-*O*-[(*R*)-2-hydroxy-1-methylethyl]-D-glucose residue. The above data indicated the presence of the disaccharide sequence 3 in the carboxyl-reduced polysaccharide.



Sugar analysis of the second saccharide gave L-altrose + 1,6-anhydro-L-altrose and erythritol in equimolar amounts. The 1H NMR spectrum showed only one anomeric proton signal at δ 5.05 ($J_{1,2} < 2$ Hz), corresponding to the β -L-altrose residue. These data indicated the presence of a second disaccharide sequence (4) in the carboxyl-reduced polysaccharide.



Additional sequence information on the carboxyl-reduced polysaccharide was obtained by inter-residue NOEs between the anomeric proton of one residue and the proton on the linkage carbon of the next residue. The inter-residue NOEs obtained for all disaccharide elements in the repeating unit of the polysaccharide are shown in Table 3. A NOE connectivity was observed between H-1 of the 4-substituted D-glucose residue and H-3 of the 3-substituted 4-*O*-[(*R*)-2-hydroxy-1-methylethyl]-D-glucose residue, and between H-1 of the 4-substituted 6-*O*-[(*R*)-2-hydroxy-1-methylethyl]-D-galactose residue and H-4 of the 2,4-disubstituted L-altrose residue. These data indicated that the compounds 3 and 4, obtained from the Smith degradation, originated from the backbone structure of the carboxyl-reduced polysaccharide, and showed the backbone structure of the carboxyl-reduced polysaccharide to be the following (5):

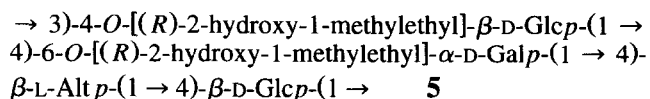


Table 3

Observed NOEs in the NOESY spectrum for anomeric protons in the carboxyl-reduced polysaccharide from *Butyrivibrio fibrisolvens* strain CF3

H-1 δ ; residue	NOE δ ; residue, proton
5.23; $\rightarrow 2,4$)- β -L-Alt p-(1 \rightarrow (A)	4.15; A, H-2 3.62; C, H-4 3.35; A, H-5
5.08; $\rightarrow 4$)-[6- <i>O</i> -R]- α -D-Galp-(1 \rightarrow ^a (B)	3.92; A, H-4 ^b 3.92; B, H-2
4.89; $\rightarrow 4$)- β -D-Glcp-(1 \rightarrow (C)	3.86; D, H-3 3.69; C, H-3 3.56; D, H-2
4.66; $\rightarrow 3$)-[4- <i>O</i> -R]- β -D-Glcp-(1 \rightarrow ^a (D)	4.23; B, H-4 3.86; D, H-3 3.53; D, H-5
4.55; β -D-Glcp-(1 \rightarrow (E)	4.15; A, H-2 3.50; E, H-3 3.45; E, H-5

^a R = (*R*)-2-hydroxy-1-methylethyl.

^b Coincides with the intra-residual NOE of H-2.

tuted in a strain-specific way. Thus an α -L-Rhap substitution is found at O-2 of α -D-Galp in strain X6C61, an *O*-acetyl group is located at O-3 of α -D-Galp in strain 49, and a β -D-Glcp residue is attached to O-2 of β -L-Alt p in strain CF3. This creates wide divergencies in the resulting structures, even though such structures may have evolved from a

common precursor epitope. Lending some additional support for this concept is the observation that in all three strains studied to date, the α -D-Galp or β -L-Alt p residue is in turn linked to O-4 of a β -D-Glcp residue, which is the next sugar in the polysaccharide chain.

The polysaccharide of *B. fibrisolvens* strain CF3 also contains 4-*O*-[(*R*)-1-carboxyethyl]-D-glucose, another rather unusual monosaccharide which has been previously identified in the capsular polysaccharide of *Klebsiella* serotype 66 [10] and in the type 3 O-antigen of *Shigella dysenteriae* [23]. The corresponding diastereomer of this sugar has been identified in the capsular polysaccharide of *Aerococcus viridans* var *homari* [13]. This sugar has not been found in the polysaccharides of either *B. fibrisolvens* strain 49 or X6C61, but is apparently present in related strains isolated by Lewis and Dehority [24] as we have previously reported [1].

3. Experimental

General methods.—Concentrations were performed under reduced pressure at 40 °C, or by flushing with N₂. For GLC, a Hewlett–Packard 5890 instrument fitted with a flame-ionization detector was used. Separation of the alditol acetates and the partially methylated alditol acetates was performed on an HP-5 fused silica capillary column, using a temperature program from 140 °C (3 min) to 240 °C at 3 °C/min. GLC–MS was performed on a Hewlett–

Table 4

Observed ²J_{C,H} and ³J_{C,H} connectivities from anomeric carbons and protons, as observed in a ¹H-detected HMBC experiment of the carboxyl-reduced polysaccharide from *Butyrivibrio fibrisolvens* strain CF3

J _{C,H} connectivities to ¹ H atom δ ; residue, hydrogen	Anomeric carbon (δ , ppm)	Residue	Anomeric proton (δ , ppm)	J _{C,H} connectivities to ¹³ C atom δ ; residue, carbon
4.48; A, H-3 3.62; C, H-4	99.77	$\rightarrow 2,4$)- β -L-Alt p-(1 \rightarrow (A)	5.23	79.10; C, C-4 77.10; A, C-2
–	101.69	$\rightarrow 4$)-[6- <i>O</i> -R]- α -D-Galp-(1 \rightarrow ^a (B)	5.08	70.88; B, C-3
3.86; D, H-3 3.36; C, H-2	103.31	$\rightarrow 4$)- β -D-Glcp-(1 \rightarrow (C)	4.89	82.57; D, C-3
4.23; B, H-4 3.56; D, H-2	104.41	$\rightarrow 3$)-[4- <i>O</i> -R]- β -D-Glcp-(1 \rightarrow ^a (D)	4.66	79.68; B, C-4
4.15; A, H-2 3.34; E, H-2	102.14	β -D-Glcp-(1 \rightarrow (E)	4.55	77.10; A, C-2

^a R = (*R*)-2-hydroxy-1-methylethyl.

Packard 5970 MSD instrument, using the column and conditions mentioned above. Methylation analysis was performed as described [25] and the methylated products were recovered using a Sep-Pak C-18 cartridge [26]. NMR spectra were recorded for solns in D₂O at 70 °C, using a Varian VXR 400 instrument. The native polysaccharide samples were passed through a column of Dowex 50 (Na⁺-form) prior to NMR analysis. ¹H and ¹³C chemical shifts are reported in ppm, using sodium 3-trimethylsilylpropionate-*d*₄ (TSP, δ_{H} 0.00) and acetone (δ_{C} 31.07), respectively, as internal references. 2D (COSY, relay and double relay COSY, NOESY, HMBC, and HMQC) experiments were performed according to standard pulse sequences available in the Varian software. A 90° pulse was used in the correlation experiments, mixing times of 0.3 and 0.6 s in the NOESY experiments, and a delay time of 60 ms in the HMBC experiments.

Pyridine–acetate buffer (0.1 M, pH 6.8) was used as mobile phase for all gel filtrations. The eluate was monitored using a Knauer differential refractometer and all fractions were checked by ¹H NMR spectroscopy.

Organism, growth conditions, and isolation of extracellular polysaccharides.—*B. fibrisolvens* strain CF3, originally isolated by Lewis and Dehority [24], was grown on 1% glucose at 39 °C under anaerobic conditions on the defined medium of Cotta and Hespell [27] supplemented with 0.3% Trypticase. Cells were removed from stationary-phase cultures by centrifugation, and crude extracellular polysaccharide was phenol-extracted, dialyzed, and lyophilized as previously described [2]. The crude polysaccharide was further purified by anion-exchange chromatography. In a typical run, crude polysaccharide (200 mg) dissolved in H₂O (200 mL) was applied to a column (1.5 × 12 cm) of DEAE-Sephadex 6-B. The column was eluted first with water (200 mL) and then with a linear gradient of aq NaCl (0–1 M, 260 mL). Aliquots (0.2 mL) of the fractions were analyzed for carbohydrate using the phenol/H₂SO₄ procedure, and the tubes containing the polysaccharide were pooled, dialyzed against H₂O, and recovered after freeze-drying. Typically, this procedure yielded 80–90 mg of purified acidic polysaccharide.

Isolation of the unknown (1-carboxyethyl)hexose.—Polysaccharide (20 mg) was hydrolyzed with 2 M CF₃CO₂H (2 mL) for 1 h at 120 °C; the hydrolysate was freeze-dried, redissolved in H₂O (1 mL), and the soln was applied to a small column (1 × 2 cm) of Dowex 1X8-50 (OH[−]-form). After elution of the

neutral sugars with H₂O (5 mL), the acidic sugars were eluted with 1 M NH₄OH (5 mL). The ammonia soln was freeze-dried, redissolved in H₂O (1 mL), and the soln was applied to a column (1.6 × 70 cm) of Dowex 50W-X8 (200–400 mesh; Ca²⁺-form). The column was irrigated with H₂O, and the eluent was monitored with a differential refractometer. The fractions were analyzed by ¹H NMR spectroscopy. Fraction 1 contained only the unknown (1-carboxyethyl)hexose (1.5 mg). Part of this product was de-alkylated with BBr₃ using a modification of the procedure described by Hough and Theobald [12]. (1-Carboxyethyl)hexose (0.2 mg) was suspended under N₂ in CH₂Cl₂ (0.75 mL), cooled to −70 °C, and 16% (v/v) BBr₃ in CH₂Cl₂ (0.25 mL, −70 °C) prepared under N₂, was added. The mixture was allowed to react in a Teflon-capped sealed tube for 30 min at −70 °C, and then permitted to gradually regain room temperature over a period of 10 h. The solvent and excess reagents were evaporated under a stream of N₂, the residue redissolved in MeOH (0.3 mL), and then dried with a stream of N₂. The washed, dried residue was treated with 10% (v/v) CF₃CO₂H (0.25 mL, 100 °C, 1 h), to afford a hexose that was identified by GLC and GLC–MS of its alditol acetate derivative. Its absolute configuration was determined according to Gerwig et al. [8].

Carboxyl reduction of the native polysaccharide.—Reduction was performed according to Taylor et al. [7] using 1-ethyl-3-(3-dimethylaminopropyl)carbodiimide (EDC) and either NaBH₄ or NaBD₄ as the reducing agent. The procedure was repeated twice in order to obtain a more complete reduction. The reaction products were checked by ¹H NMR spectroscopy to monitor the reduction efficiency.

Determination of the configuration of 4-O-[(R)-2-hydroxy-1-methylethyl]-D-glucose.—The acetate of 4-O-(2-hydroxy-1-methylethyl-2,2-*d*₂)-D-glucose, obtained from the carboxyl-reduced (NaBD₄) polysaccharide [7], was compared by GLC–MS with the acetate of 4-O-[(S)-2-hydroxy-1-methylethyl]-D-glucose prepared from the carboxyl-reduced [7] extracellular polysaccharide from *Aerococcus viridans* var *homari* [13] using the same conditions as in the sugar analysis.

Smith degradation.—Carboxyl-reduced (NaBH₄) polysaccharide (27 mg) was treated with a soln of NaIO₄ (67 mg) in NaOAc buffer (0.1 M, pH 3.7; 2.5 mL) for 20 h at 4 °C. Excess NaIO₄ was destroyed by addition of ethylene glycol (50 μ L), and the product was reduced with NaBH₄ (30 mg in 3.0 mL 1 M NH₄OH), dialyzed against distilled water, and then

freeze-dried. A ^1H NMR spectrum of the product indicated incomplete oxidation, so the above procedure was repeated to obtain a more completely oxidized product (21 mg). The polymer was treated with 0.5 M $\text{CF}_3\text{CO}_2\text{H}$ (0.5 mL) at room temperature for 24 h, then the soln was freeze-dried, and the products were separated by gel filtration on a column (1.5×70 cm) of Bio-Gel P-2. The fractions were analyzed by ^1H NMR spectroscopy. A portion of each saccharide (≈ 0.3 mg) was hydrolyzed in 2 M $\text{CF}_3\text{CO}_2\text{H}$ for 1 h at 120°C , the released sugars were reduced with NaBH_4 and acetylated with 1:1 Ac_2O –pyridine (0.5 mL) for 1 h at 120°C . The resulting alditol acetates were analyzed by GLC–MS.

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References

- [1] R.J. Stack, *Appl. Environ. Microbiol.*, 54 (1988) 878–883.
- [2] R.J. Stack, *FEMS Microbiol. Lett.*, 48 (1987) 83–87.
- [3] M. Andersson, S. Ratnayake, L. Kenne, L. Ericsson, and R.J. Stack, *Carbohydr. Res.*, 246 (1993) 291–301.
- [4] R.J. Stack and D. Weisleder, *Biochem. J.*, 268 (1990) 281–285.
- [5] F. Ferreira, M. Andersson, L. Kenne, M.A. Cotta, and R.J. Stack, *Carbohydr. Res.*, 278 (1995) 143–153.
- [6] R.J. Stack, R.D. Plattner, and G.L. Cote, *FEMS Microbiol. Lett.*, 51 (1988) 1–6.
- [7] R.L. Taylor, J.E. Shively, and H.E. Conrad, *Methods Carbohydr. Chem.*, 7 (1976) 149–151.
- [8] G.J. Gerwig, J.P. Kamerling, and J.F.G. Vliegenthart, *Carbohydr. Res.*, 77 (1979) 1–7.
- [9] M. Andersson, L. Kenne, R. Stenutz, and G. Widmalm, *Carbohydr. Res.*, 254 (1994) 35–41.
- [10] P.-E. Jansson, B. Lindberg, J. Lönngren, and C. Ortega, *Carbohydr. Res.*, 132 (1984) 297–305.
- [11] S.J. Angyal, G.S. Bethell, and R.J. Beveridge, *Carbohydr. Res.*, 73 (1979) 9–18.
- [12] L. Hough and R.S. Theobald, *Methods Carbohydr. Chem.*, 2 (1965) 203–206.
- [13] L. Kenne, B. Lindberg, B. Lindqvist, J. Lönngren, B. Arie, R.G. Brown, and J.E. Stewart, *Carbohydr. Res.*, 51 (1976) 287–290.
- [14] S. Hakomori, *J. Biochem. (Tokyo)*, 55 (1964) 205–208.
- [15] U.R. Bhat, B.S. Krishnaiah, and R.W. Carlson, *Carbohydr. Res.*, 220 (1991) 219–227.
- [16] P.-E. Jansson, L. Kenne, and G. Widmalm, *Carbohydr. Res.*, 188 (1989) 169–191.
- [17] K. Bock and M. Beck, *Acta Chem. Scand. B*, 34 (1980) 389.
- [18] S.J. Angyal and V.A. Pickles, *Aust. J. Chem.*, 25 (1972) 1695–1710.
- [19] K. Bock and C. Pedersen, *Acta Chem. Scand. B*, 29 (1975) 258–264.
- [20] I.J. Goldstein, G.W. Hay, B.A. Lewis, and F. Smith, *Methods Carbohydr. Chem.*, 5 (1965) 361–370.
- [21] A. Haug and B. Larsen, *Carbohydr. Res.*, 17 (1971) 297–308.
- [22] M. Höök, U. Lindahl, G. Bäckström, A. Malmström, and L.-Å. Fransson, *J. Biol. Chem.*, 249 (1974) 3908–3915.
- [23] B.A. Dmitriev, V.L. Lvov, and N.K. Kochetkov, *Carbohydr. Res.*, 56 (1977) 207–209.
- [24] S.M. Lewis and B.A. Dehority, *Appl. Environ. Microbiol.*, 50 (1985) 356–363.
- [25] P.-E. Jansson, L. Kenne, H. Liedgren, B. Lindberg, and J. Lönngren, *Chem. Comm., Univ. Stockholm*, 8 (1976) 1–75.
- [26] T.J. Waeghe, A.G. Darvill, M. McNeil, and P. Albersheim, *Carbohydr. Res.*, 123 (1983) 281–304.
- [27] M.A. Cotta and R.B. Hespell, *Appl. Environ. Microbiol.*, 52 (1986) 51–58.